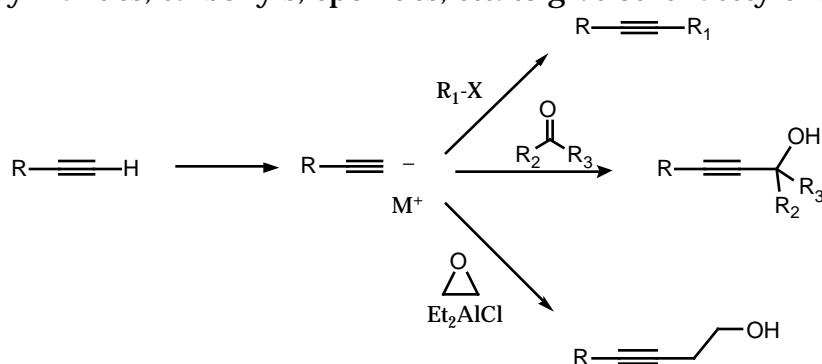


C-C Bond Formation

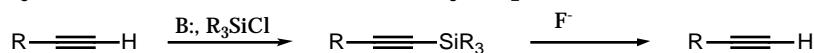
1. From other acetylenes
2. From carbonyls
3. From olefins
4. From Strained Rings
5. Eschenmosher Fragmentation
6. Allenes

From Other Acetylenes

- The proton of terminal acetylenes is acidic ($pK_a = 25$), thus they can be deprotonated to give acetylide anions which can undergo substitution reactions with alkyl halides, carbonyls, epoxides, etc. to give other acetylenes.



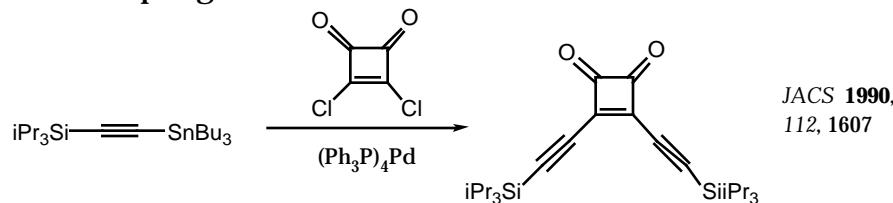
- Since the acetylenic proton is acidic, it often needs to be protected as a trialkylsilyl derivative. It is conveniently deprotected with fluoride ion.

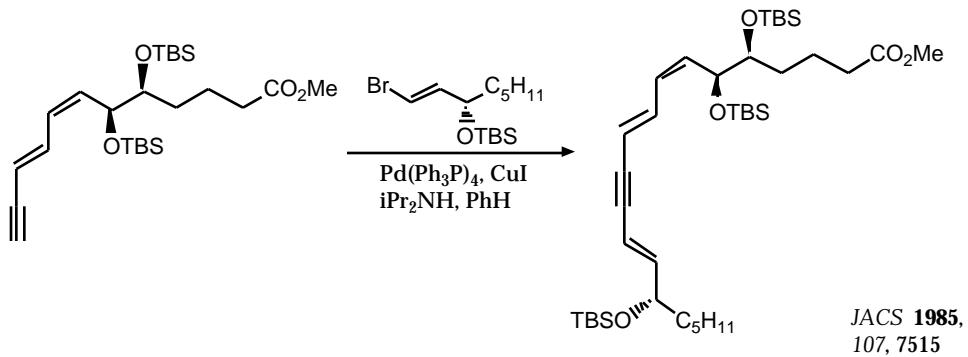


Acetylide anions and organoboranes



Palladium Coupling Reactions:



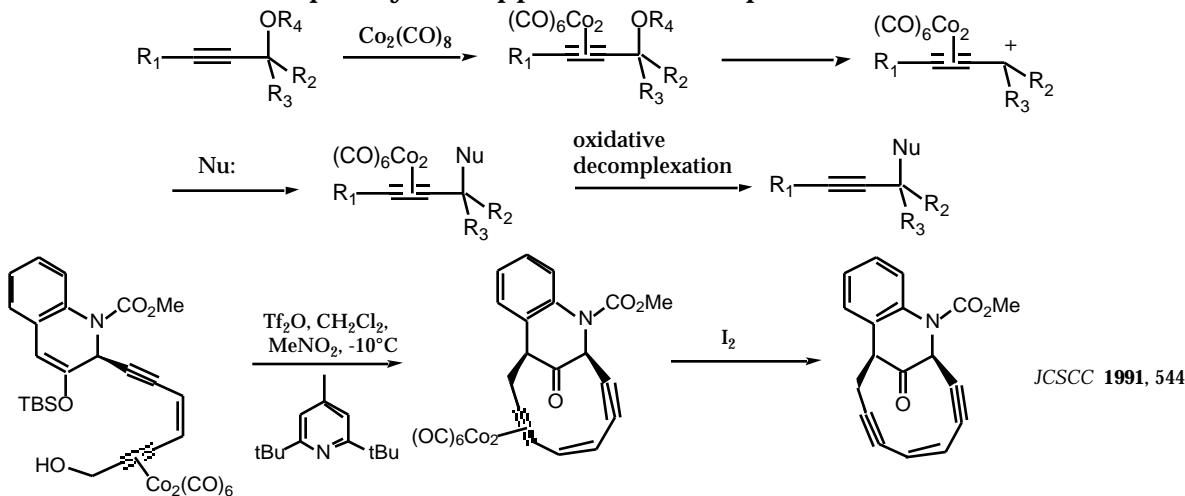


Copper Coupling- 1,3-diyynes



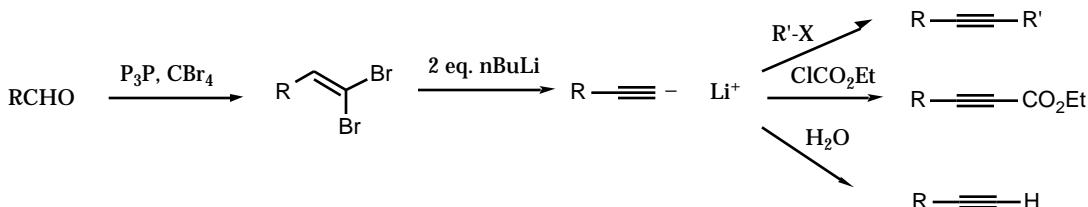
Nicholas Reaction

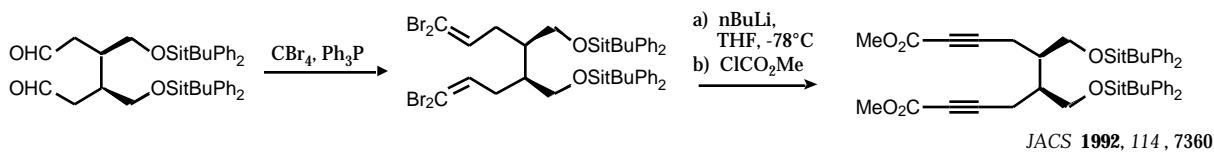
- acetylenes as their $\text{Co}_2(\text{CO})_8$ complex can stabilize an α -positive charge, which can subsequently be trapped with nucleophiles.



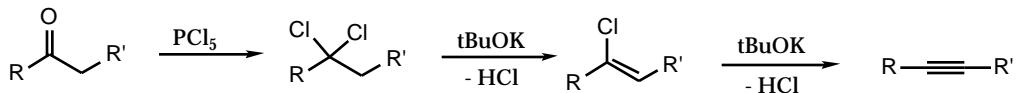
$\text{Co}_2(\text{CO})_6$ -acetylene decomplexation: *JOC* **1997**, 62, 9380

From Aldehydes and Ketones

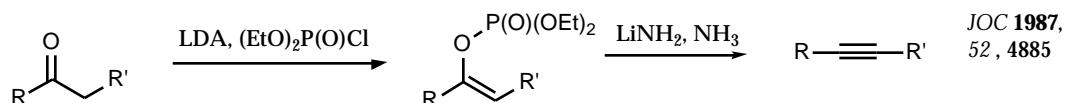




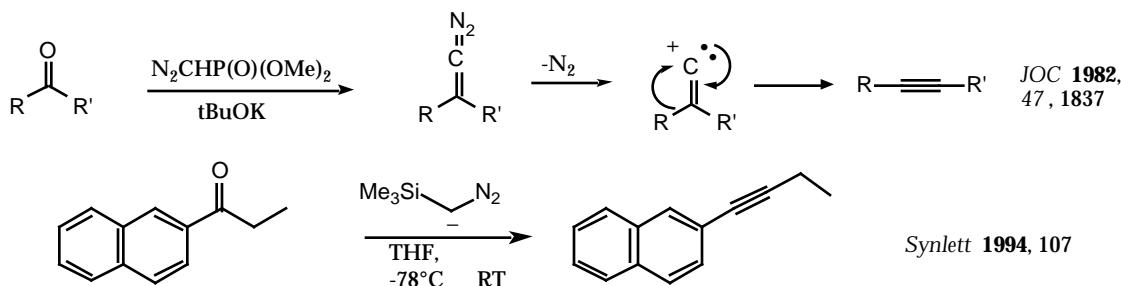
- by conversion of ketones to gem-dihalides followed by elimination



- by conversion of ketones to enol phosphates followed by elimination

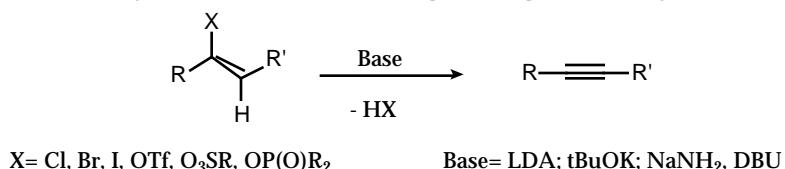


- Insertion reaction of a vinyl carbene (terminal acetylenes)

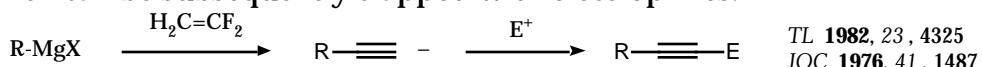


Via Elimination Reactions of Vinyl Halides

- Treatment of vinyl halides with strong base gives acetylenes.



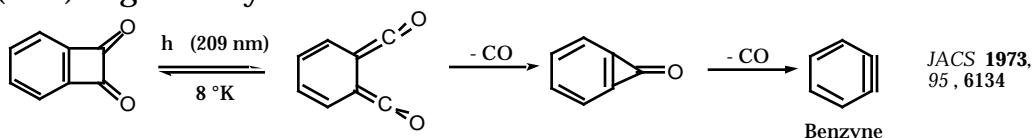
- Addition of Grignard reagents to 1,1-difluoroethylene yields an acetylidyne anion which can be subsequently trapped with electrophiles.

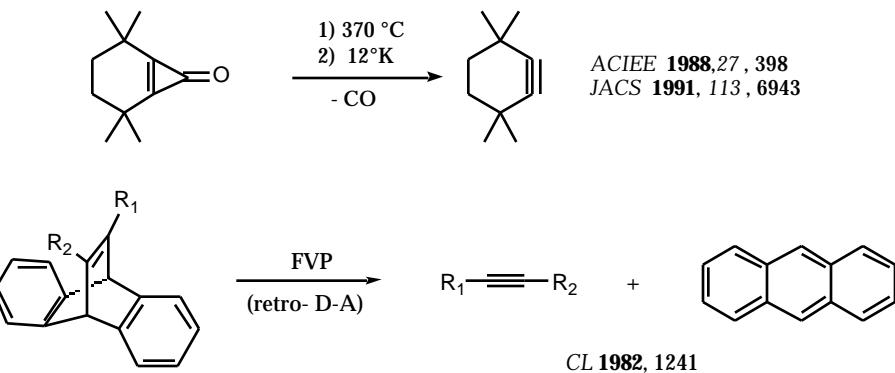
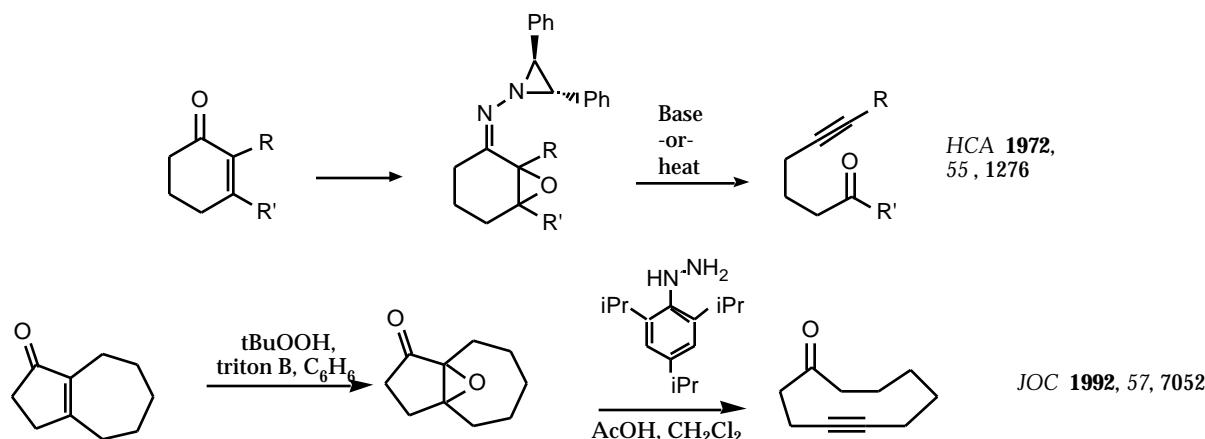
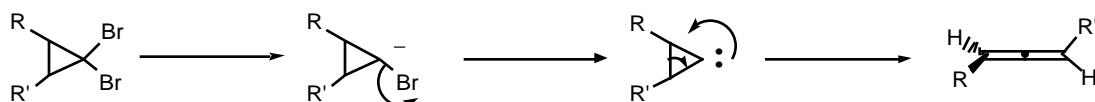


Strained Rings

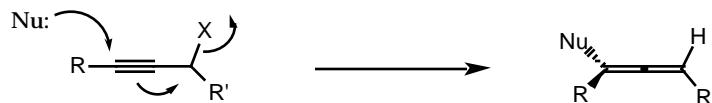
Topics in Current Chemistry 1983, 109, 189.

- Cyclopropenones and cyclobutendiones can be photolyzed or thermolyzed (FVP) to give acetylenes.



*Eschenmoser Fragmentation***Allenes** Tetrahedron 1984, 40, 2805
- from dihalocyclopropanes

- From SN2' Reactions



- from sigmatropic rearrangements from propargyl sulfoxides and phosphine oxides.

