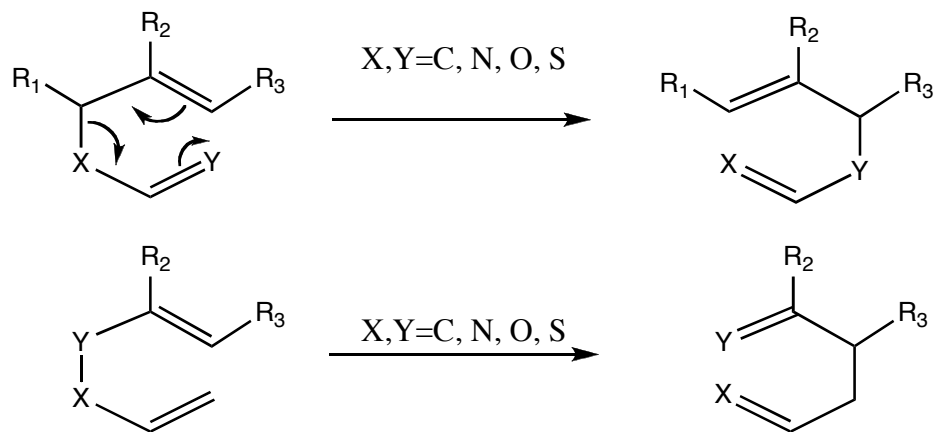
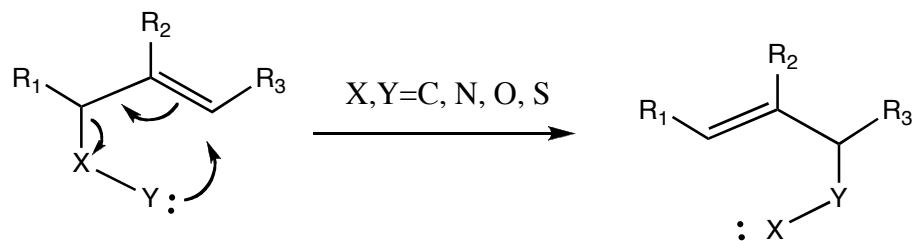


## Generalized Sigmatropic Processes

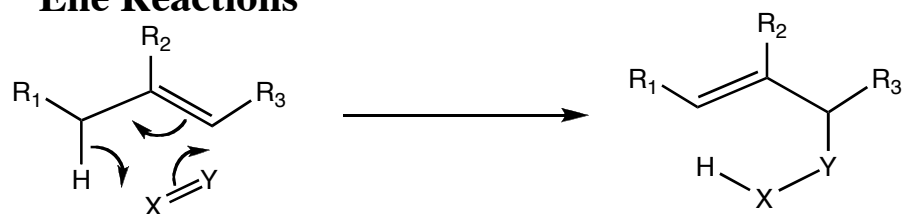
### [3,3]-sigmatropic Processes



### [2,3]-sigmatropic Processes

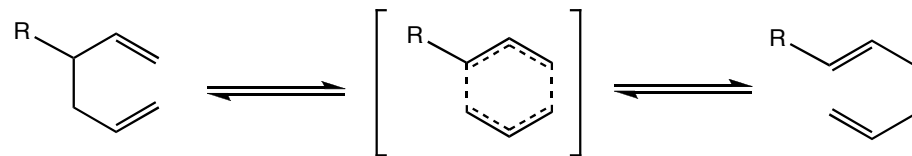


### Ene Reactions

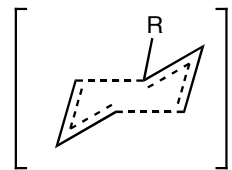


## [3,3]- Sigmatropic Rearrangements

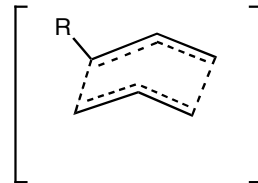
### Cope Rearrangement



Transition states:



chair

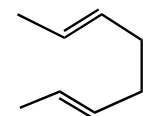
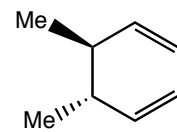


boat

(+5.3 kcal/mol)

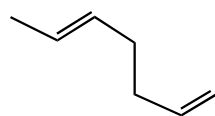
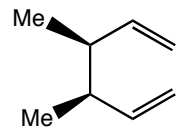
disfavored

**Rationalize:**



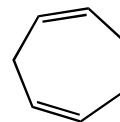
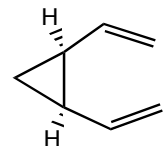
trans-trans 90%

*Tetrahedron*, **1962**, 67

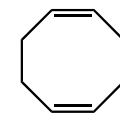
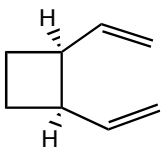


trans-cis 97%

*Ring Strain can be employed to drive the Cope process:*

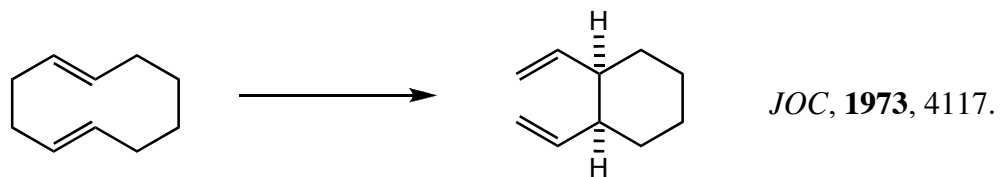


*Chem. Commun*, **1973**, 319

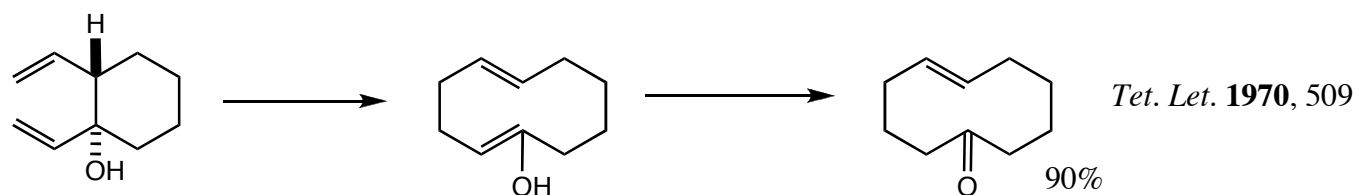


*Annalen*, **1958**, 1.

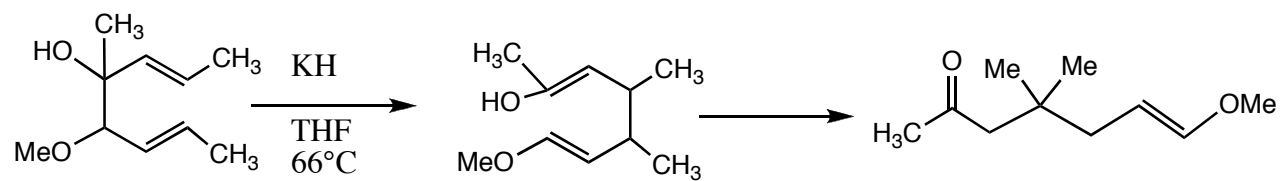
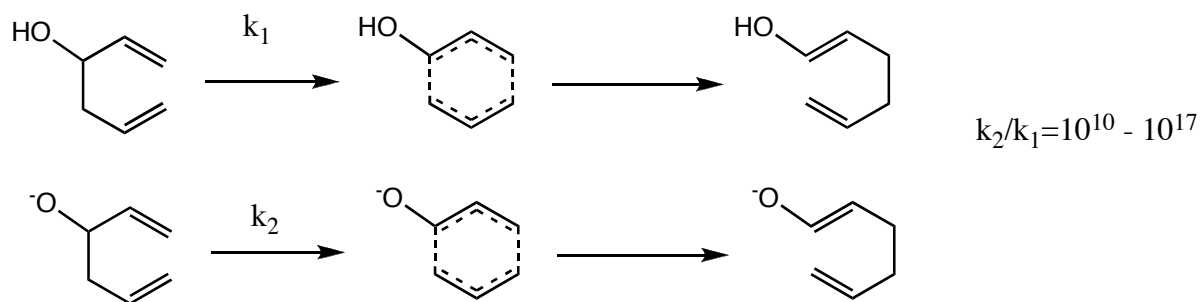
Position of Equilibrium dictated by ring strain issues:



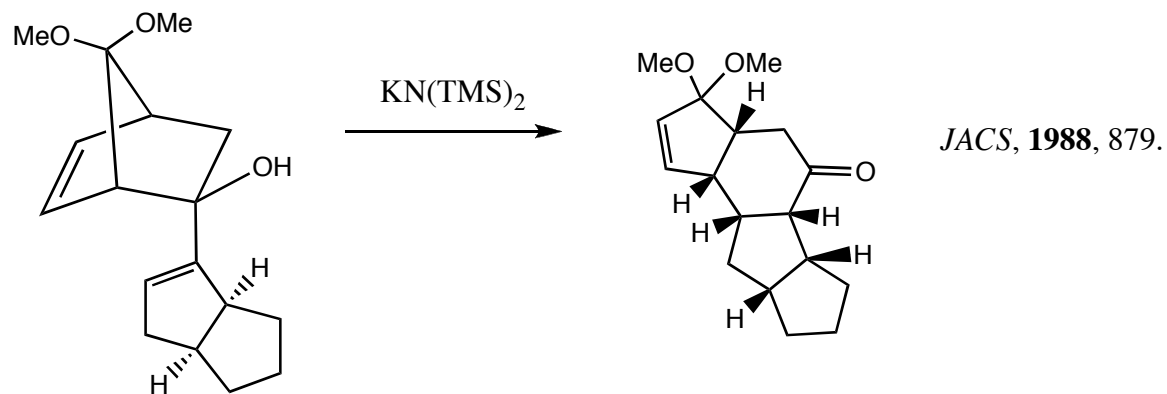
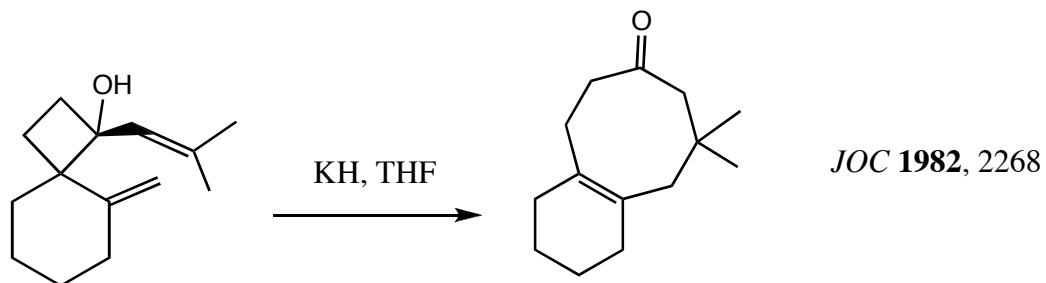
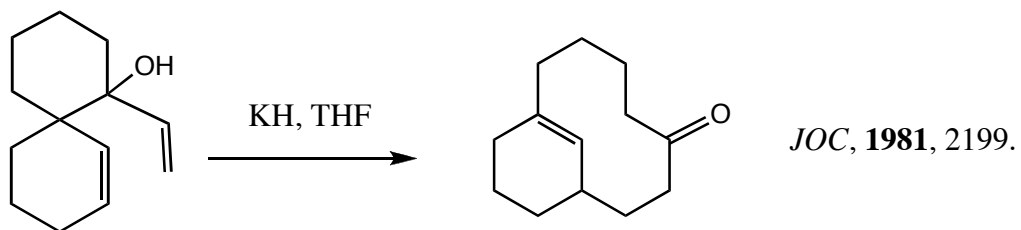
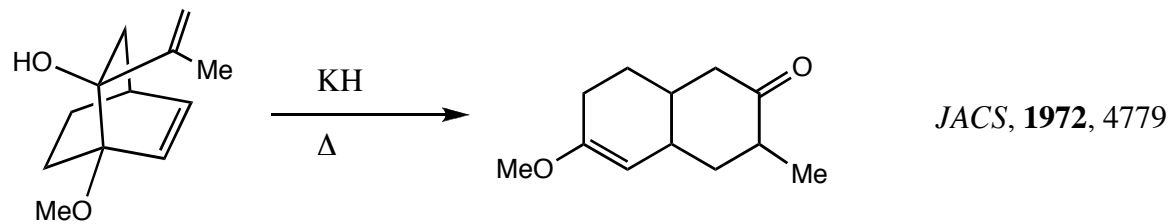
However, Tautomerism can shift the equilibrium:



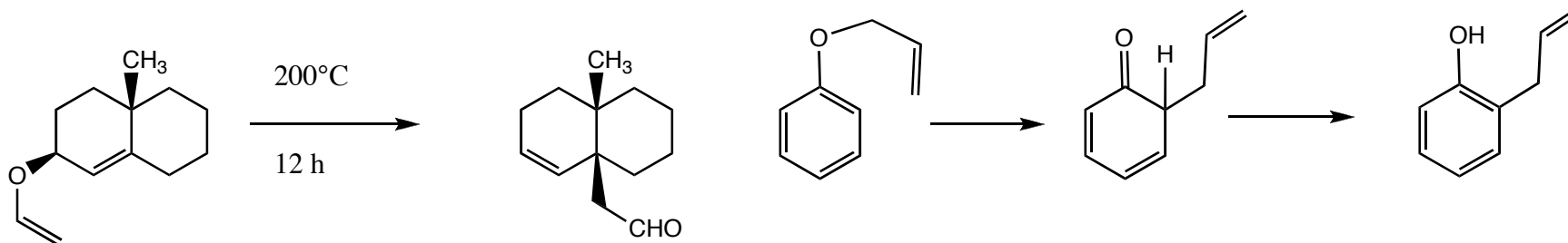
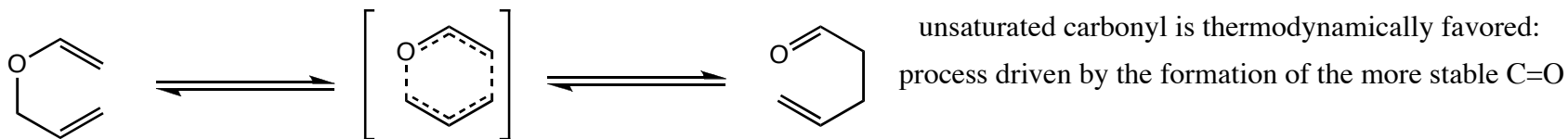
Anion-Accelerated Cope Rearrangements:



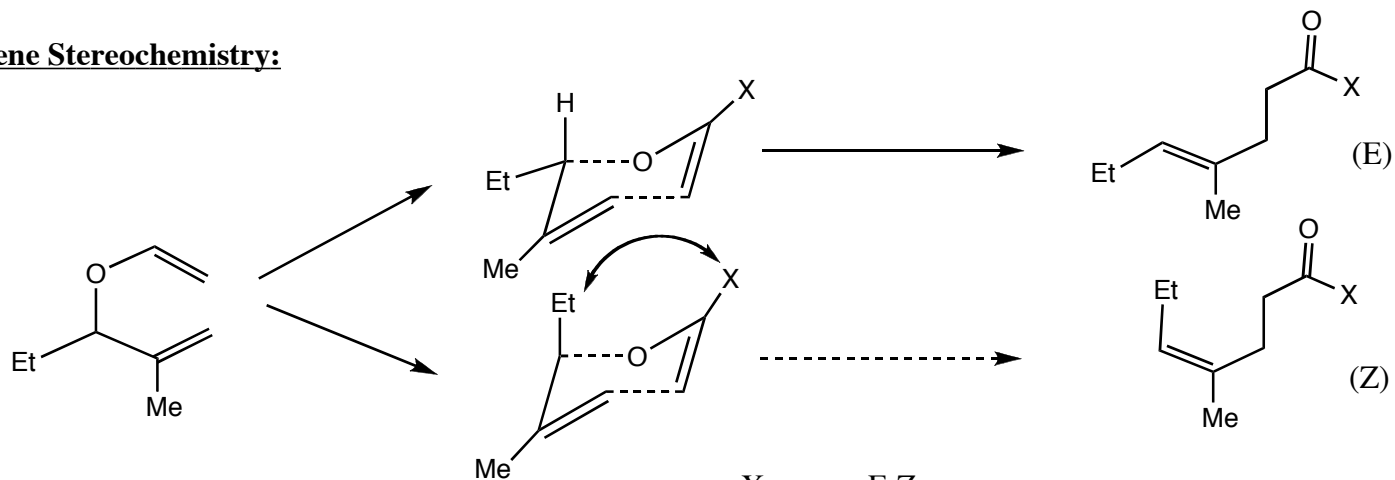
**Examples:**



## The Claisen Rearrangement



### Alkene Stereochemistry:

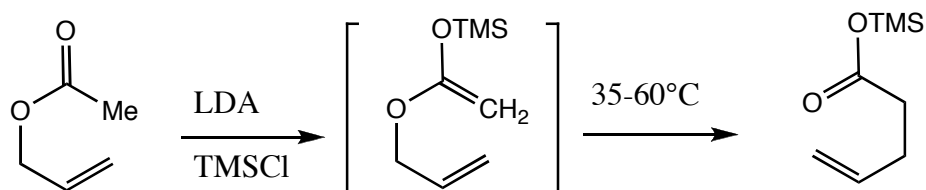


X	E:Z
H	90:10
Me	>99:1
OMe	>99:1
Me <sub>2</sub> N	>98:2

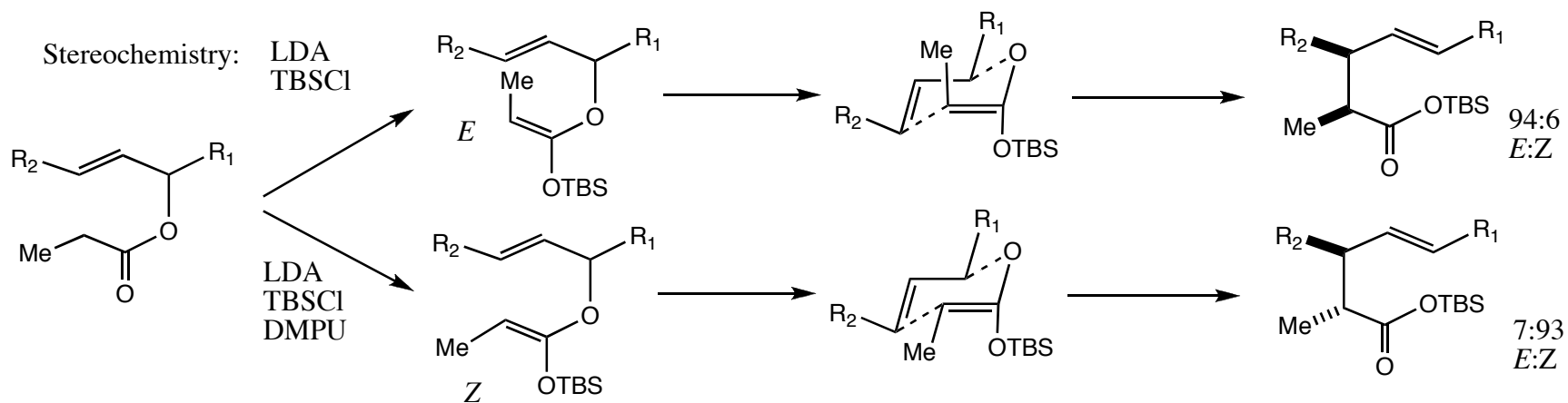
*Tet, Let.* **1969**, 3243  
*JACS*, **1973**, 553  
*JACS*, **1970**, 741

## Ireland Enolate Claisen

A low temperature, mild variant of the Claisen process:

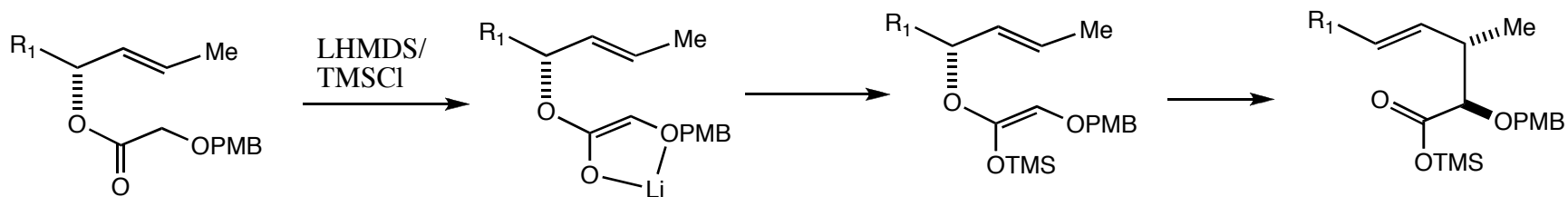


Enolates that do not fragment may be used without silyl trapping to afford the carboxylic acid products. Enolate rearrangements occur at 25°C.



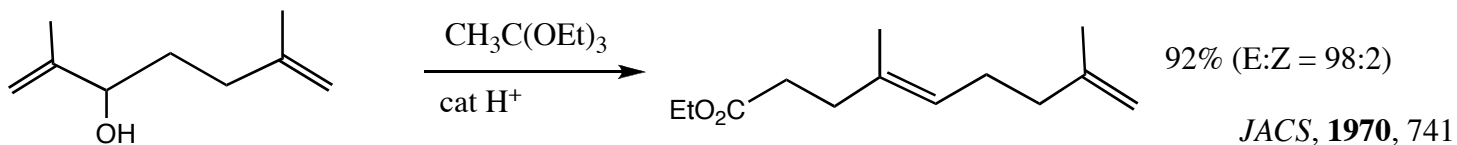
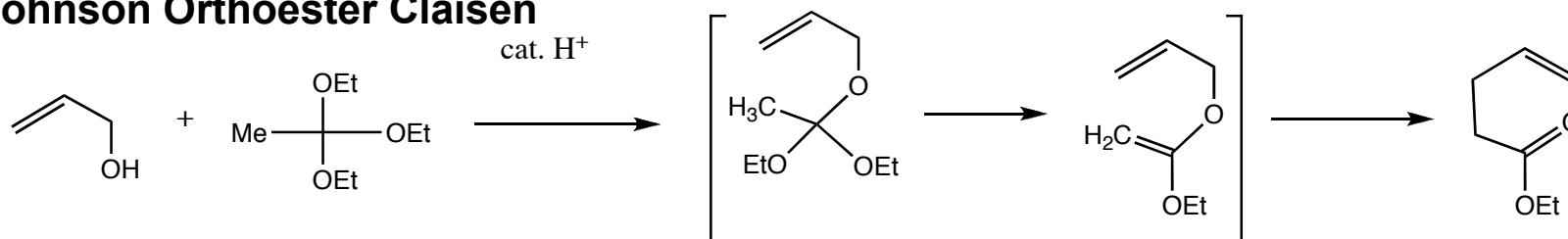
*JOC*, **1991**, 650.

*Chelating Substituents favor Z-Enolates:*

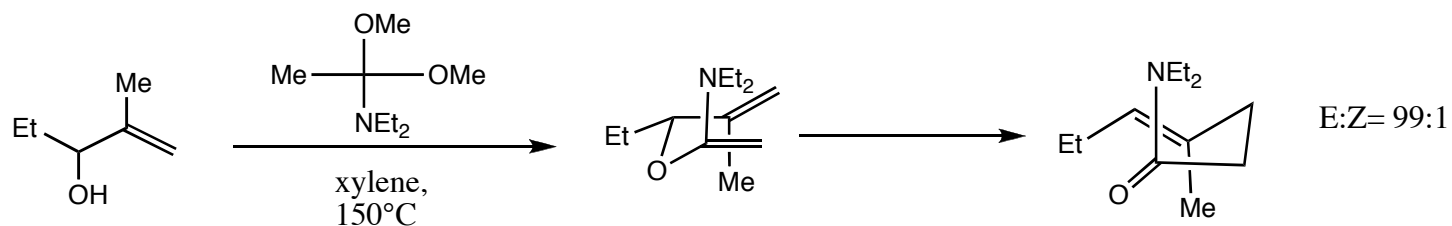
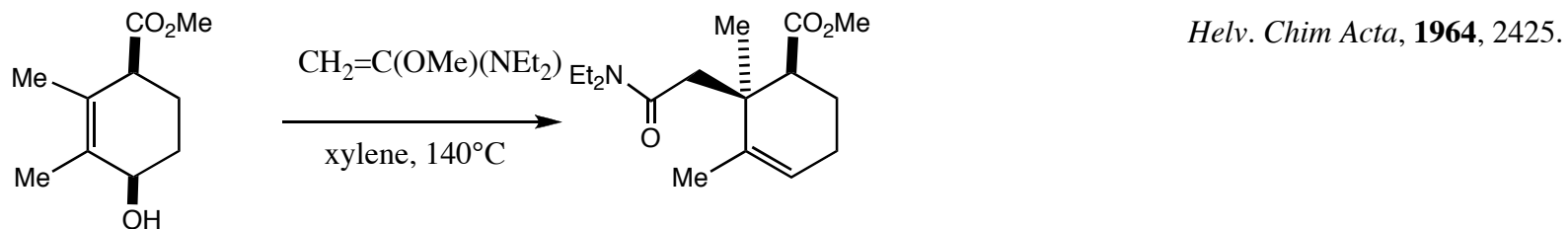
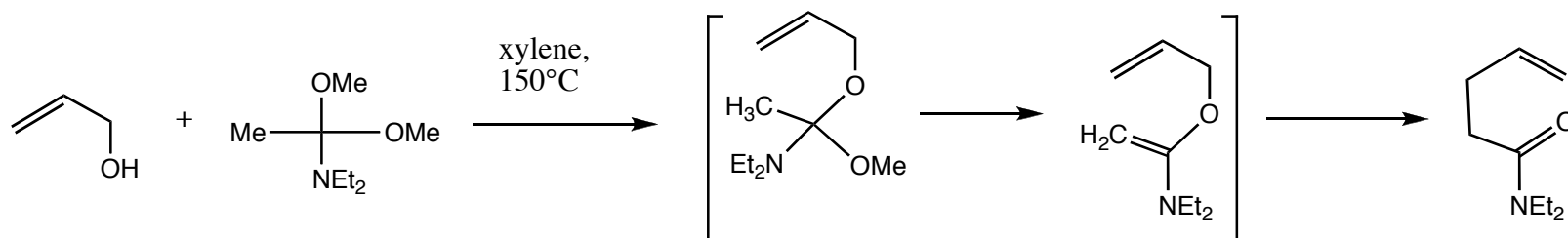


*TL* **1993**, 1103.

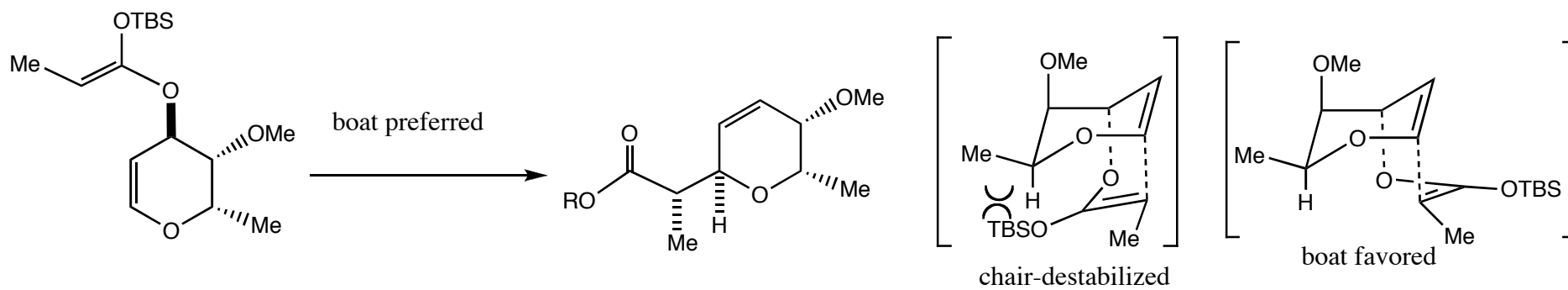
## Johnson Orthoester Claisen



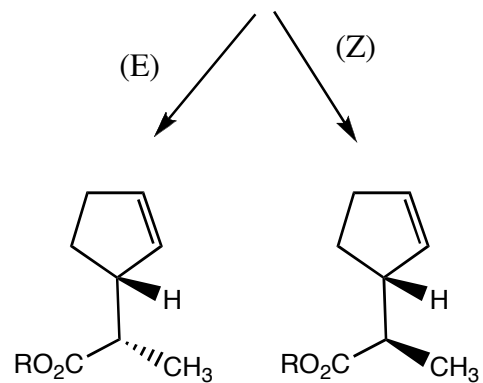
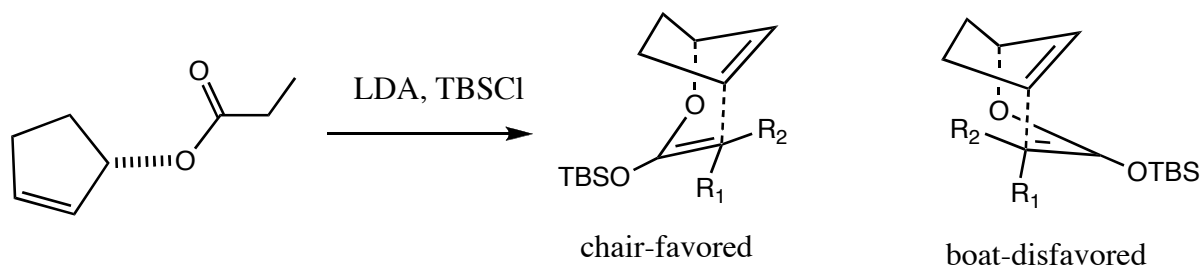
## Eschenmoser Claisen



## Boat Transition States are more accessible in Claisen than in Cope Rearrangements



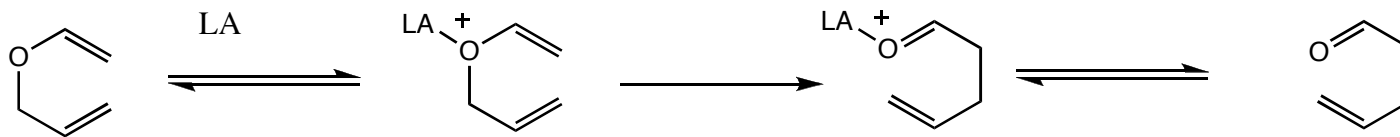
However,



*JACS*, **1991**, 3572  
*JOC*, **1981**, 3896



## Lewis Acid Catalysis of the Claisen Rearrangement

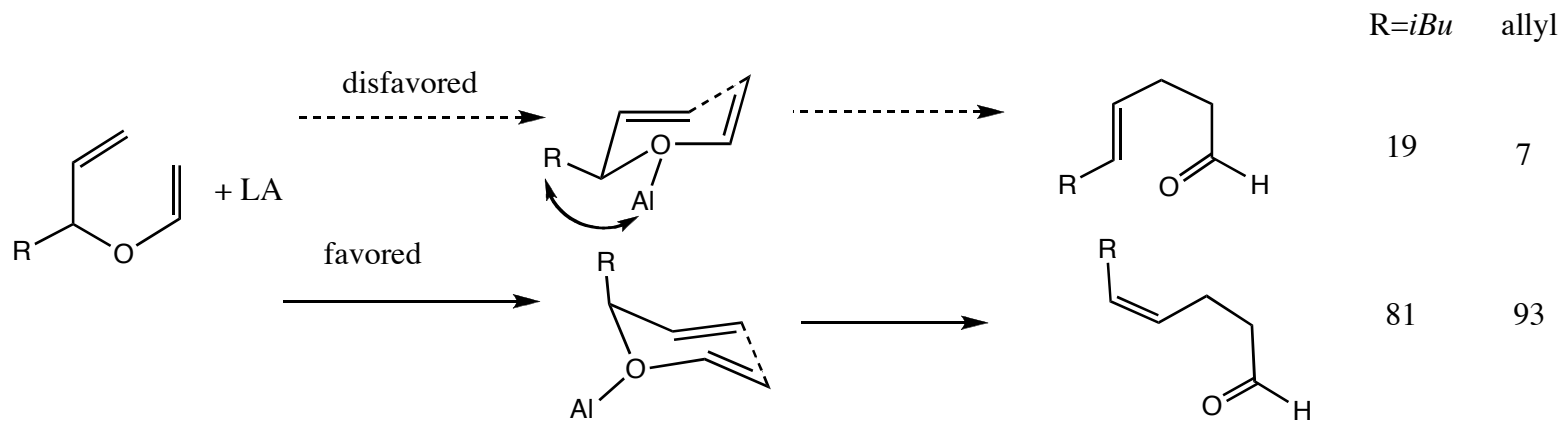


Lewis Acids:  $\text{Et}_2\text{AlCl}$  *JOC*, **1970**, 3166

$\text{TiCl}_4$ : *Chem Lett.* **1975**, 1041

$(\text{RO})_2\text{AlMe}$  *JACS*, **1988**, 7922; *TL*, **1989**, 1265; *JACS*, **1990**, 316

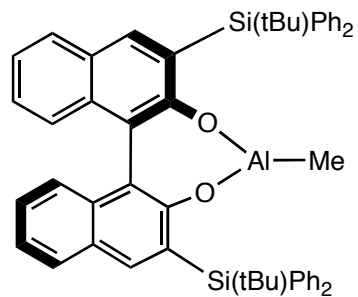
$\text{LiClO}_4$ , *Tet*, **1993**, 6025



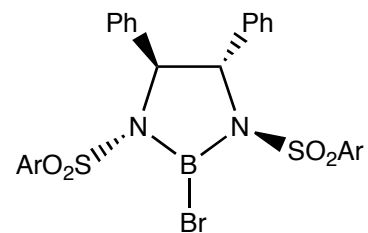
*JACS*, **1990**, 316

# Enantioselective Claisen Rearrangements

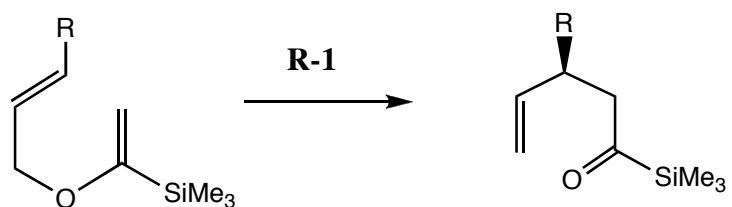
**R-1**



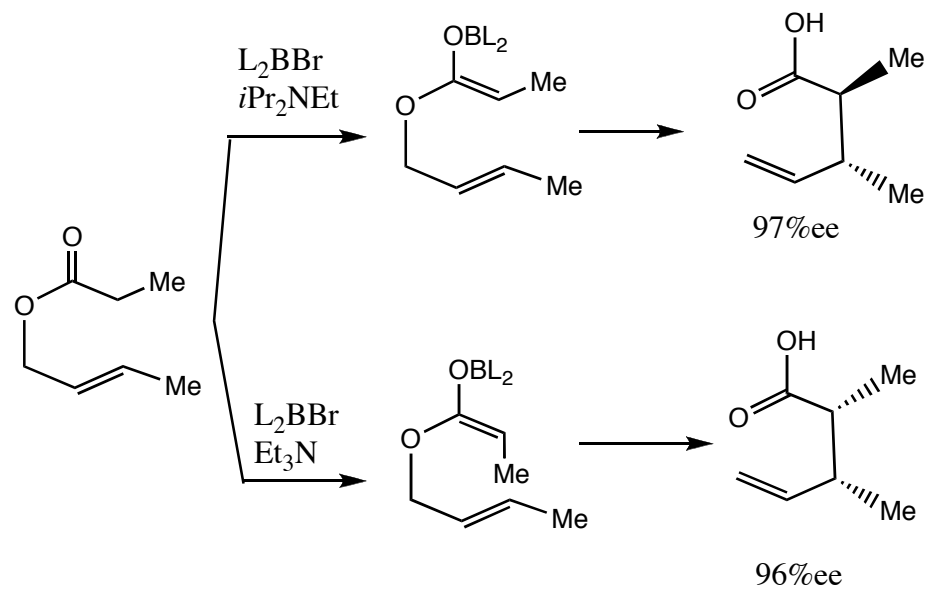
Yamamoto, *JACS*, **1990**, 7791



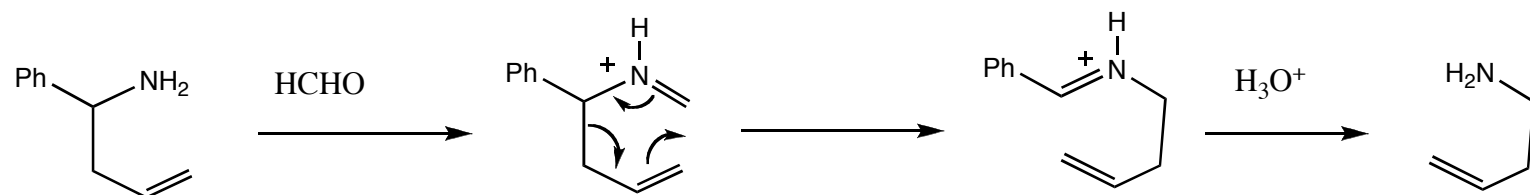
Corey, *JACS*, **1991**, 4026



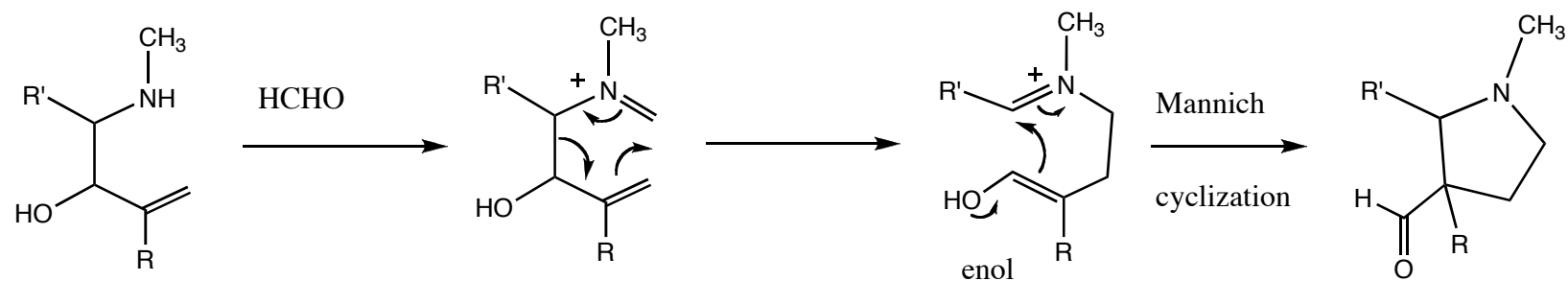
R=Ph, 88%ee  
R=C<sub>6</sub>H<sub>11</sub>, 71%ee



## The Aza-Cope Rearrangement

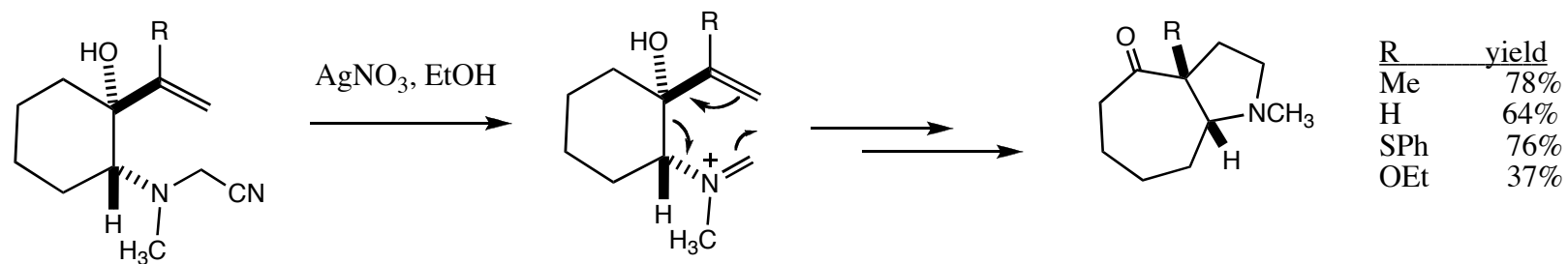


### Aza-Cope Mannich Process

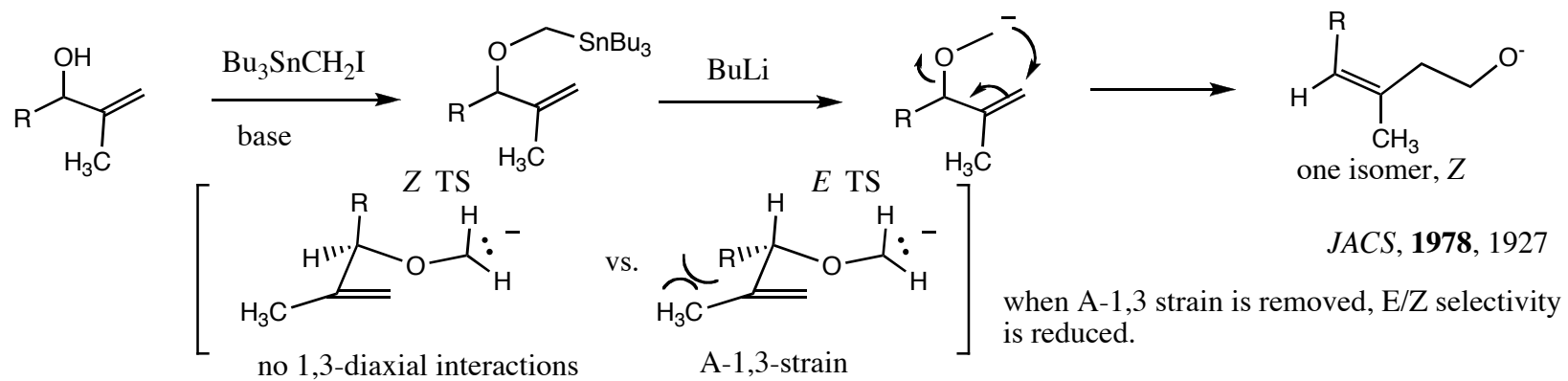
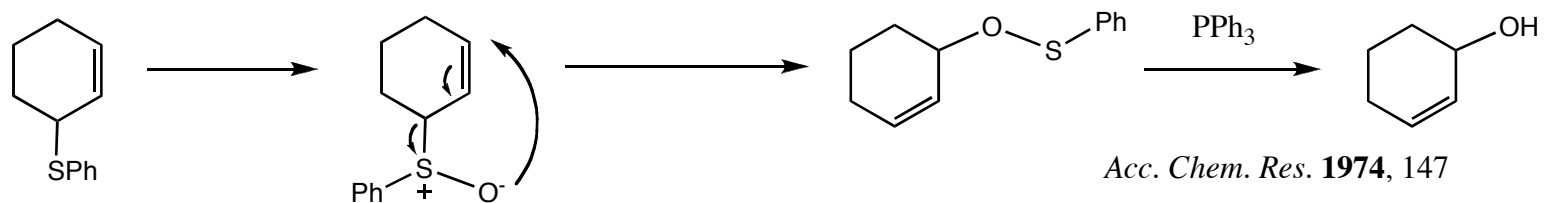
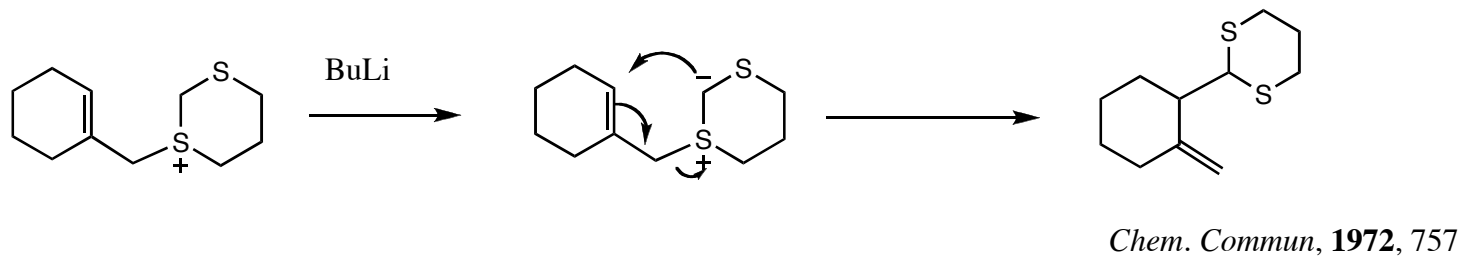
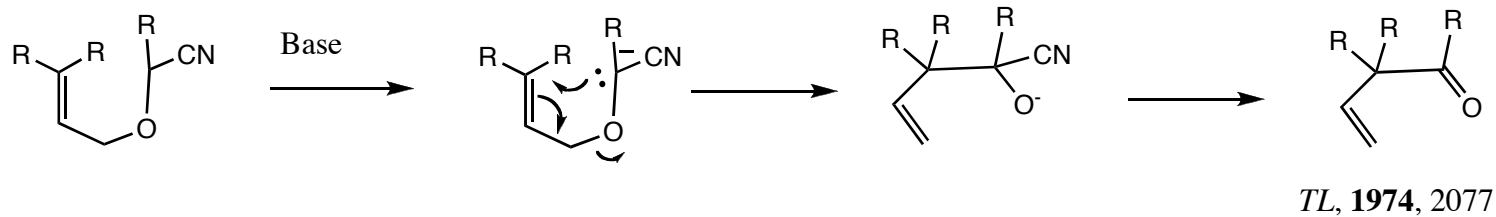


*JACS*, **1988**, 4329

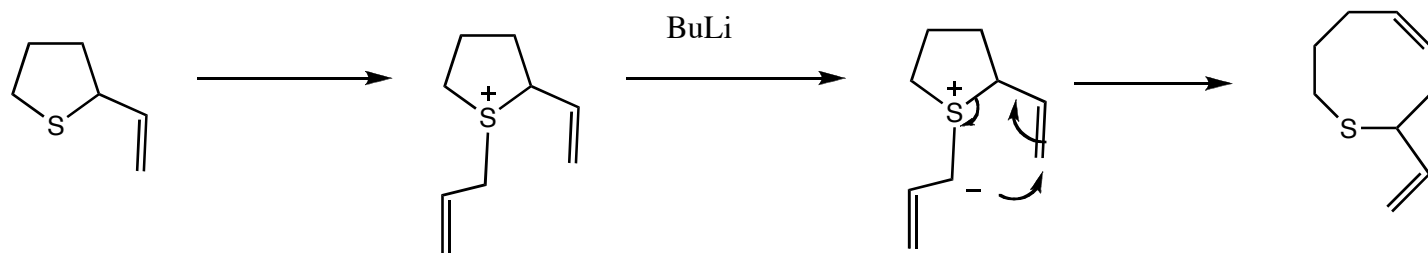
*JACS*, **1988**, 685



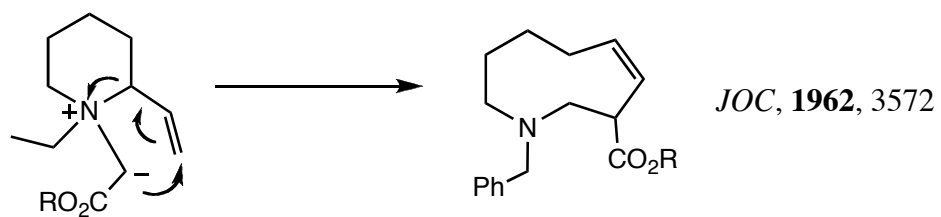
## [2,3]-Wittig Reaction



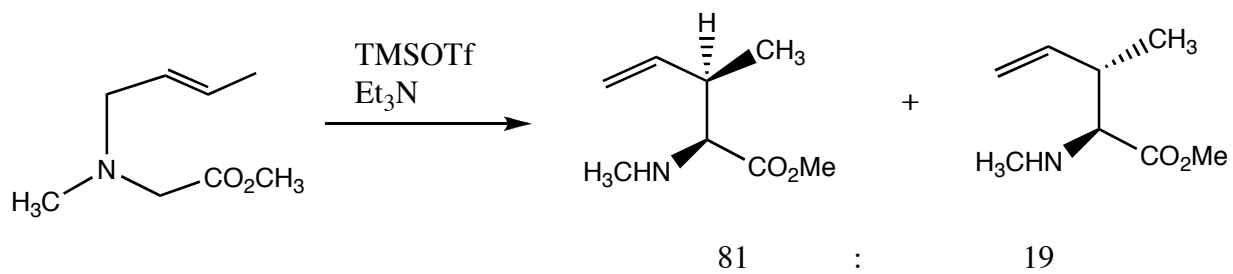
## Applications of the 2,3-Wittig in Synthesis



*JACS*, **1975**, 6878  
*JOC*, **1978**, 1185  
*TL*, **1978**, 519



*JOC*, **1962**, 3572



*Chem Lett*. **1990**, 2069.