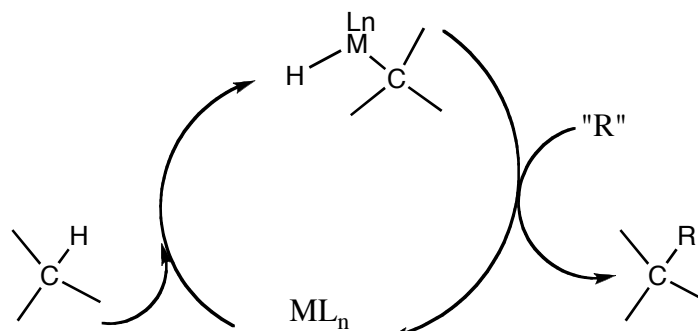
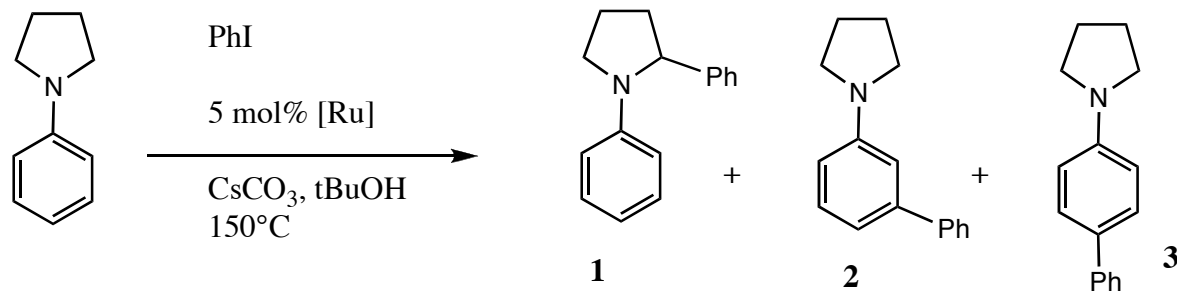


C-H activation for C-C bond formation

Direct C-H activation requires insertion of a transition metal (usually Ru, Ir, Rh or Pd) across a strong C-H bond (90-105 kcal/mol) to form a new, weaker C-M bond (50-80 kcal/mol), followed by generation of a new C-R bond



C-H activation is limited to simple aryl and vinyl sp^2 C-H bonds and C-H bonds adjacent to heteroatoms

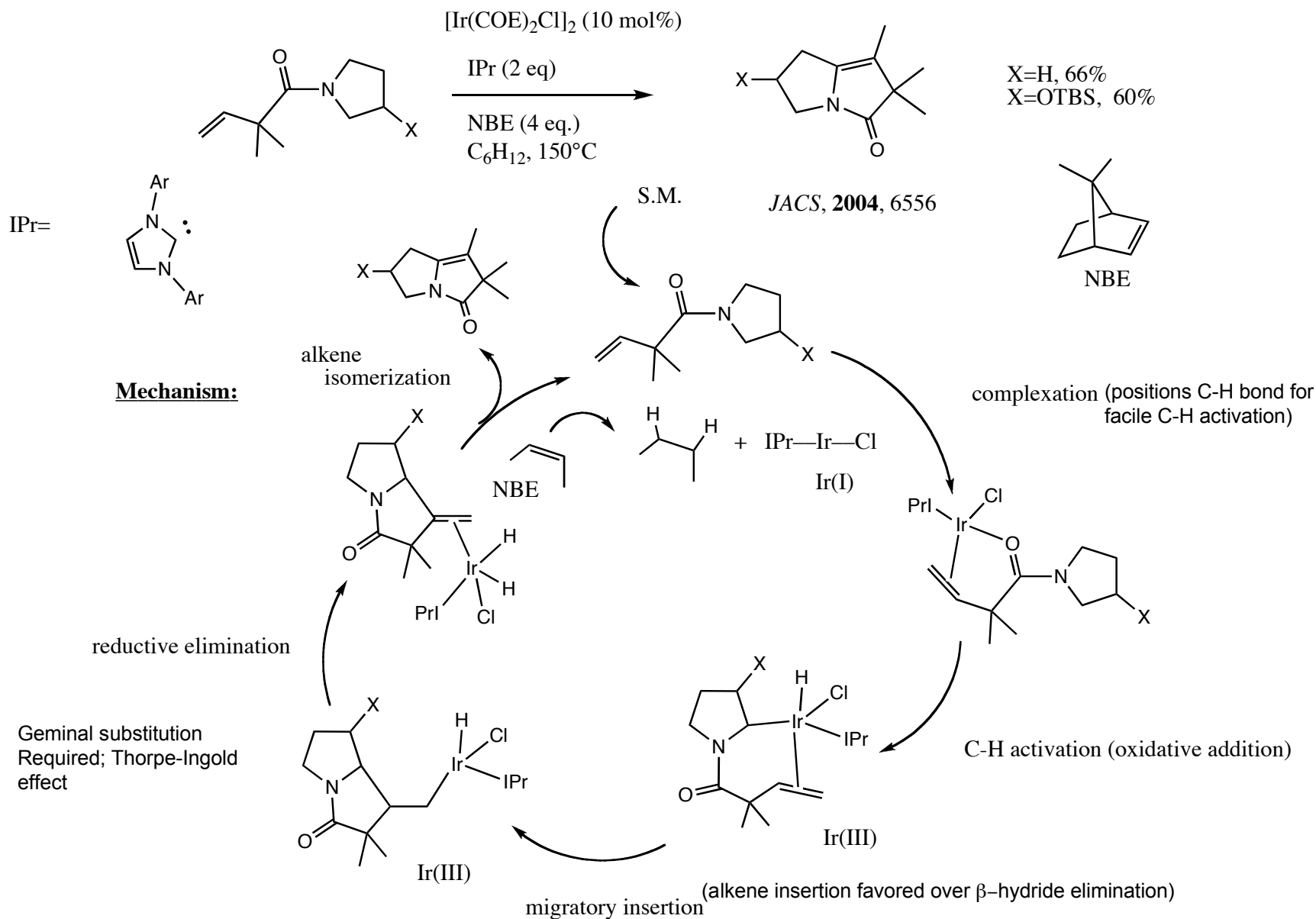


JACS, 2005, 5284

[Ru]	1	2	3	%
$Ru(H_2)_2(PCy_3)_2$	40	7	4	
$Ru(H_2)_2(H)_2(PCy_3)_2$	59	9	4	

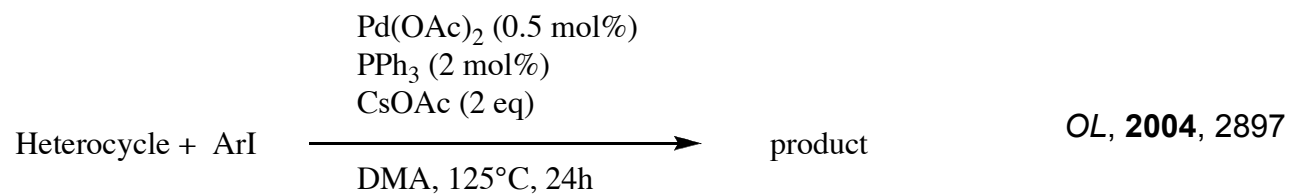
C-H activation for C-C bond formation

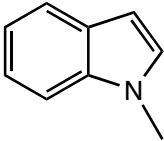
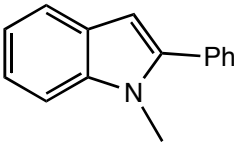
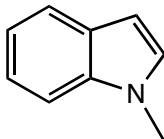
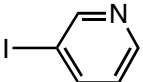
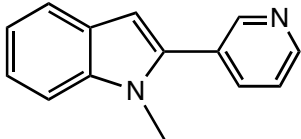
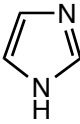
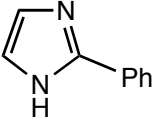
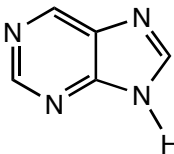
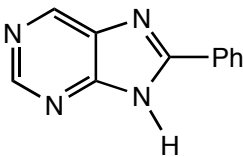
Intramolecular sp^3 C-H activation/cyclization



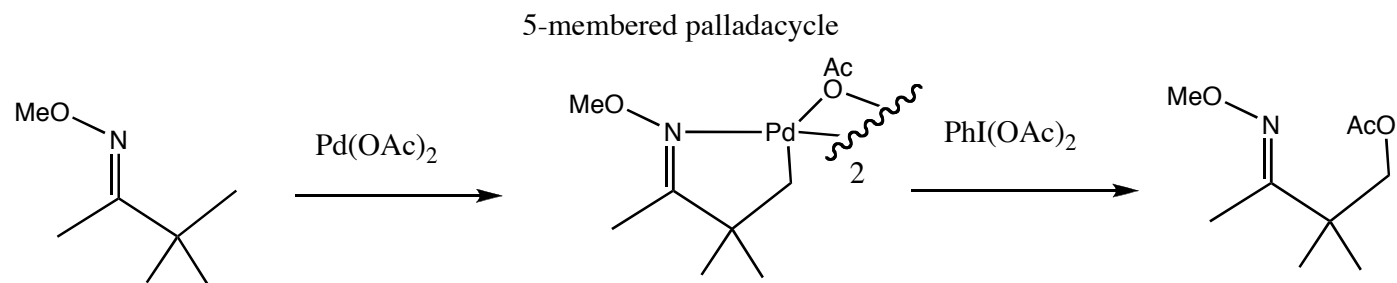
C-H activation for C-C bond formation

Activation of C(sp²)-H bonds α to N:

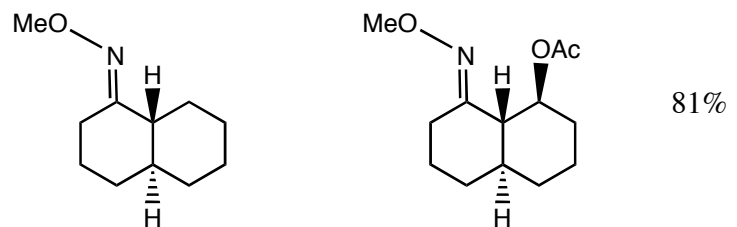
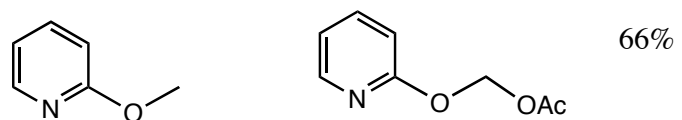
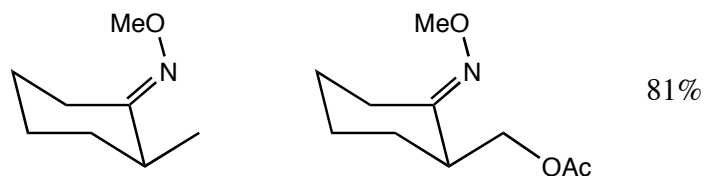
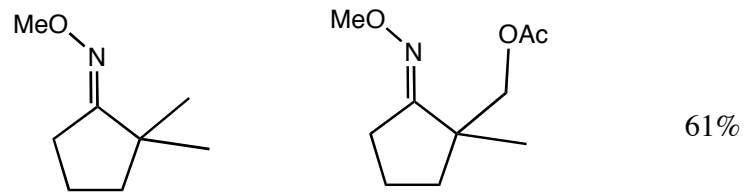


substrate	ArI	products (yield)
	Ph-I	 85%
		 73%
	Ph-I	 52%
	Ph-I	 73%

Palladium-Catalyzed Oxygenation of Unactivated C-H bonds



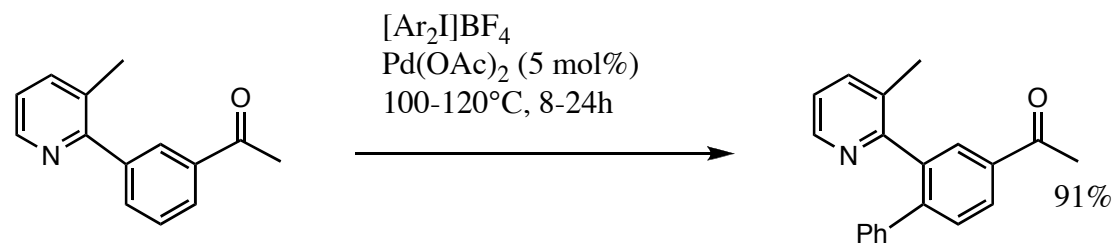
substrate **product** **yield**



JACS, 2004, 9542

C-H activation to form a palladacycle intermediate
 stoichiometric oxidant accomplishes Pd(II) \Rightarrow Pd(IV) oxidation, followed by C-O bond-forming reductive elimination

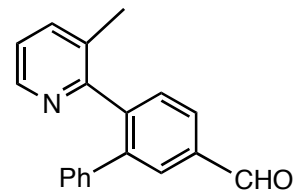
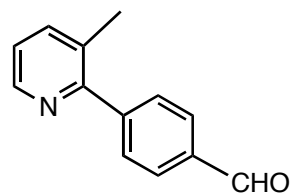
C-H activation for C-C bond formation



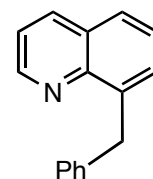
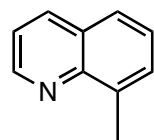
substrate

Product

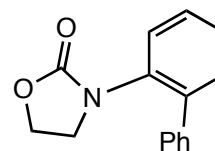
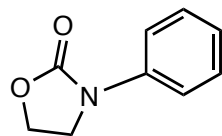
Yield



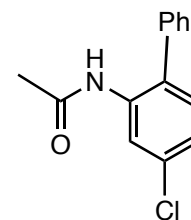
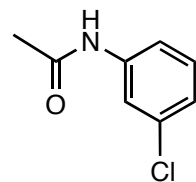
51%



72%



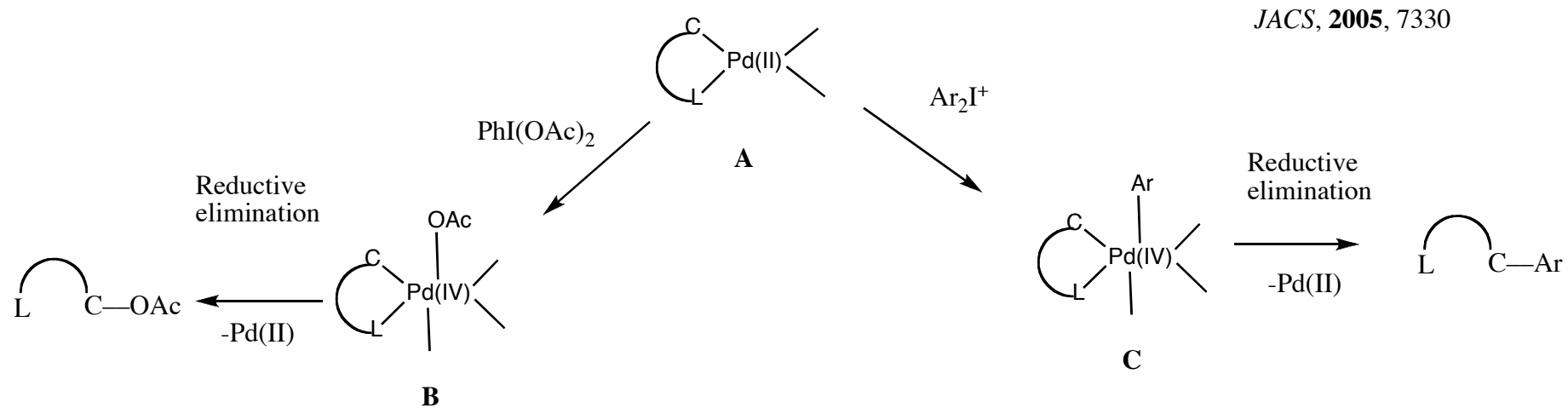
83%



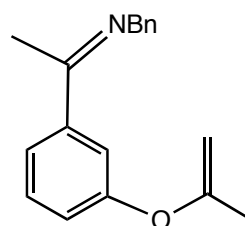
68%

Mechanism of Sanford Arylation/Oxidation: putative Pd(IV) intermediate

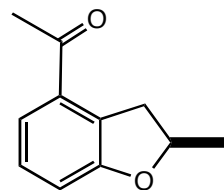
L=ligating group; C=carbon of C-H insertion site



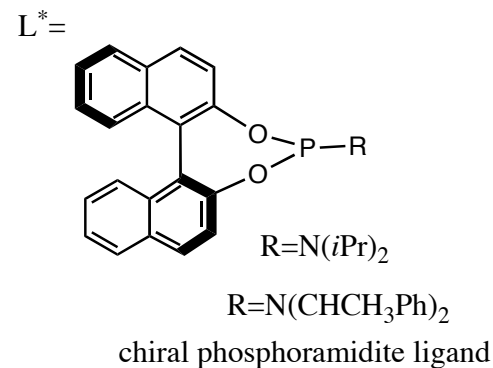
Imine-directed C-H activation and cyclization: enantioselective C-H activation



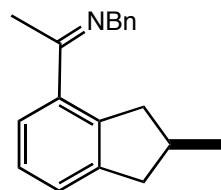
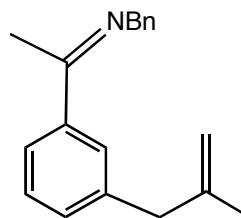
1. 5mol% $[\text{RhCl}(\text{coe})_2]_2$
15 mol% L^*
2. aqueous work-up



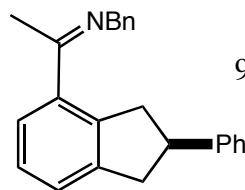
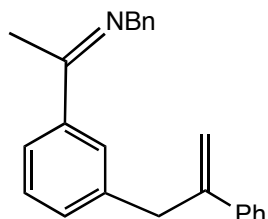
99%, 96% ee



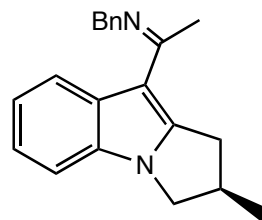
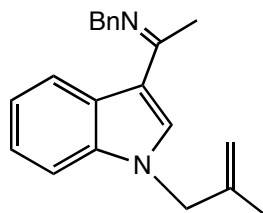
substrate **product** **yield** **ee**



94 95



96 90

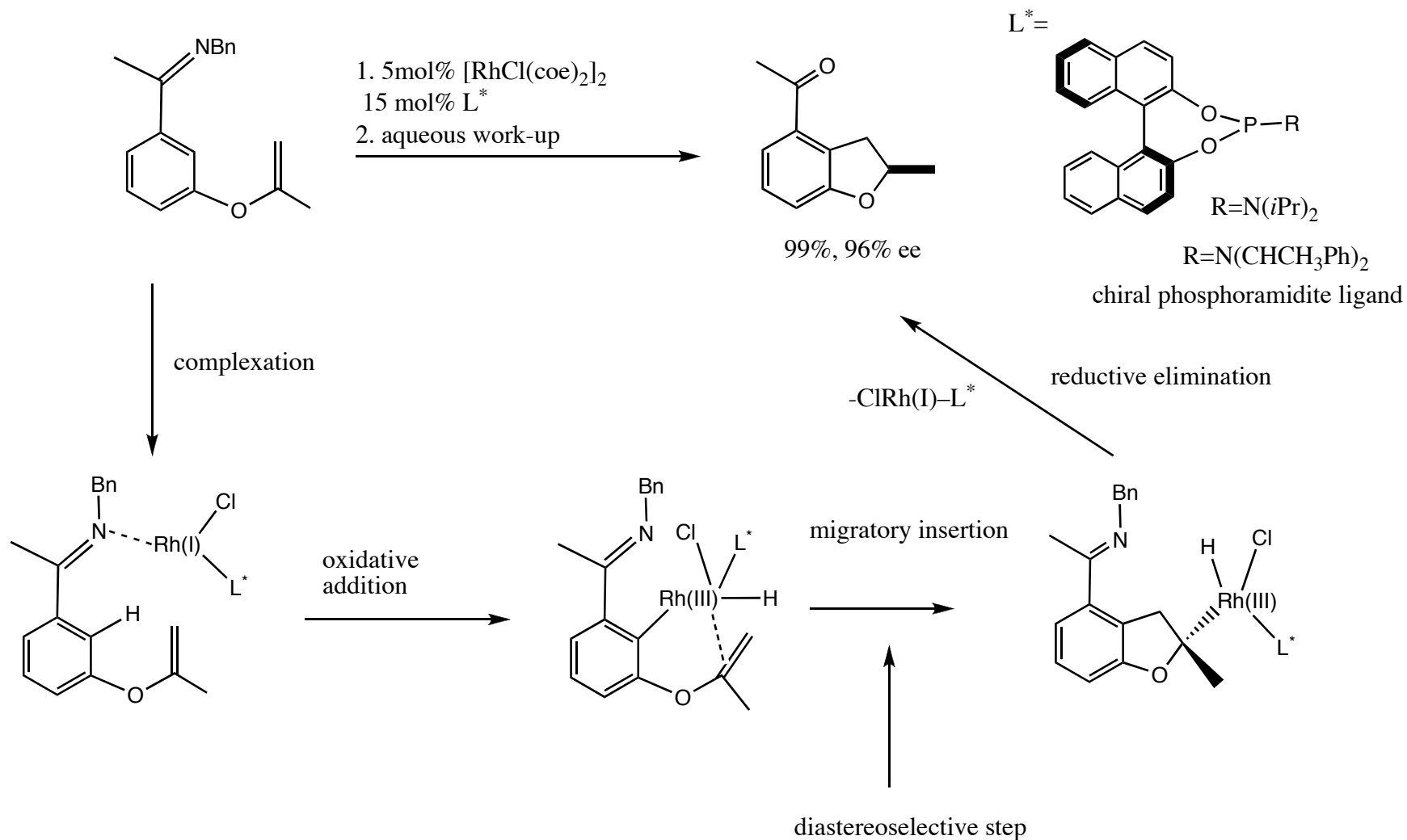


90 70

- coordination of chiral rhodium to imine
- directed C-H insertion of the α -aryl proton
- migratory insertion of the pendant olefin ligand
- reductive elimination $\text{Rh}(\text{I}) \Rightarrow \text{Rh}(\text{III}) \Rightarrow \text{Rh}(\text{I})$
- a highly diastereoselective migratory insertion of the olefin into the Rh-H bond is mediated by the chiral ligand framework

JACS, 2004, 7192

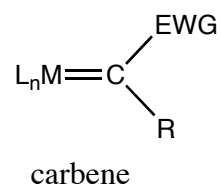
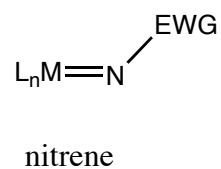
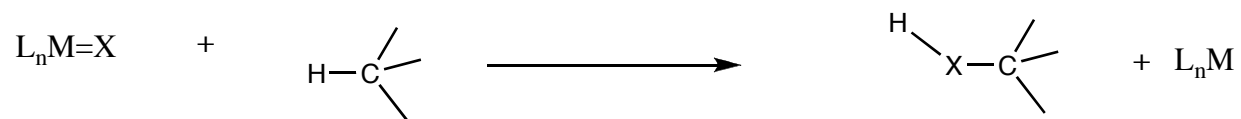
Mechanism of Bergman-Ellman Enantioselective C-H activation



Note the importance of a directing group in all C-H activation processes!

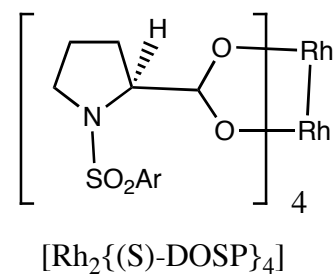
Alternative Approach to C-H activation/ C-H functionalization: use chiral carbenes

C-H insertion:

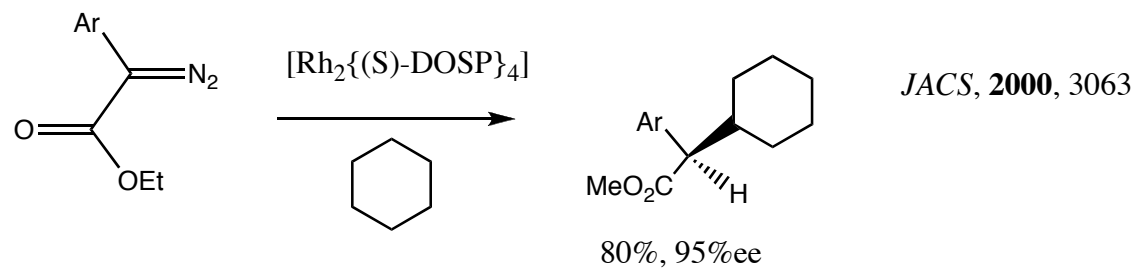


EWG required for sufficient reactivity
R=EDG increases selectivity

common chiral dirhodium for carbenoid formation:

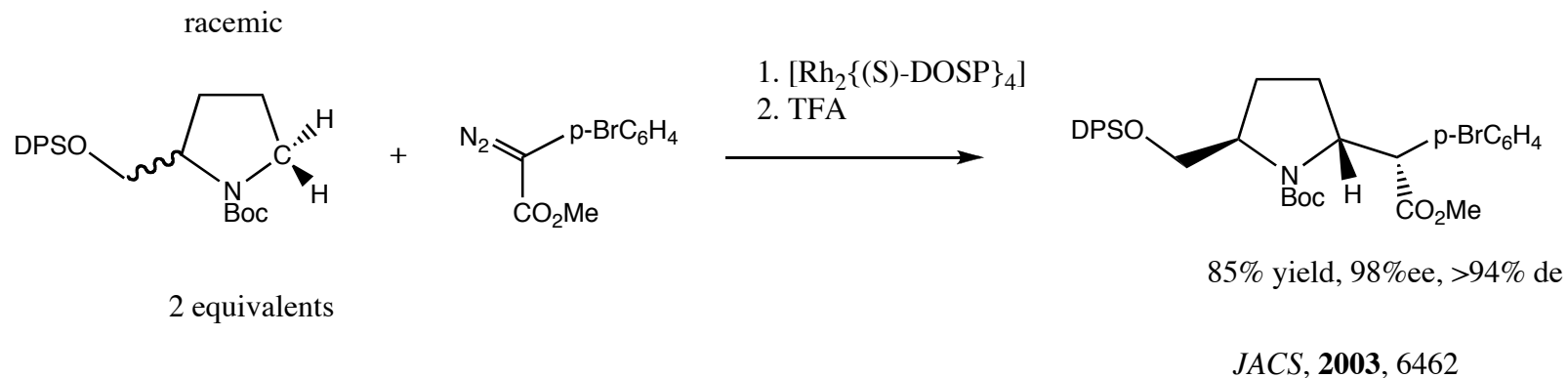


Consider:

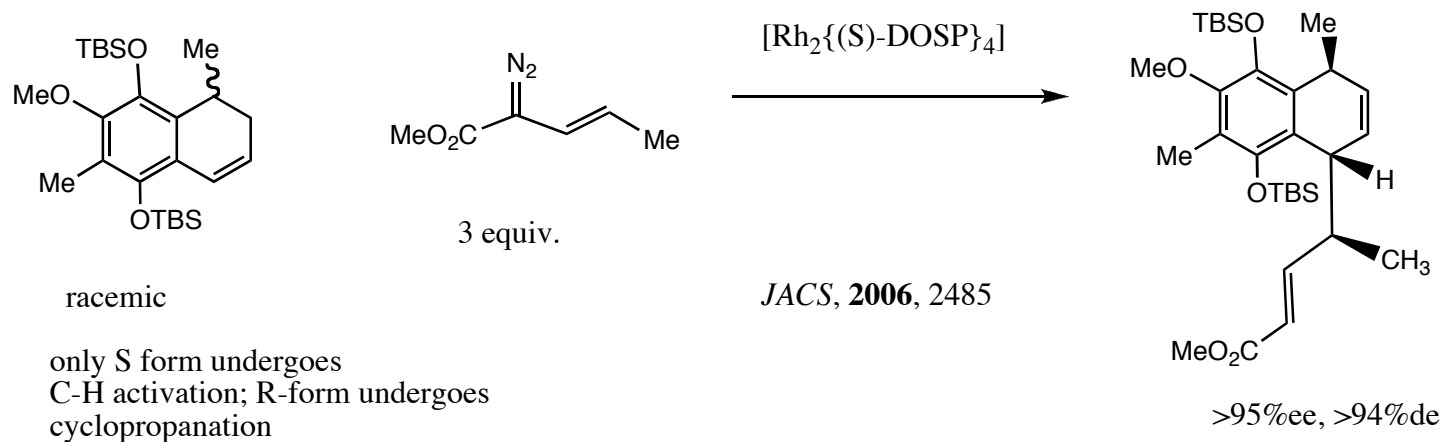


Alternative Approach to C-H activation/ C-H functionalization: use chiral carbenes

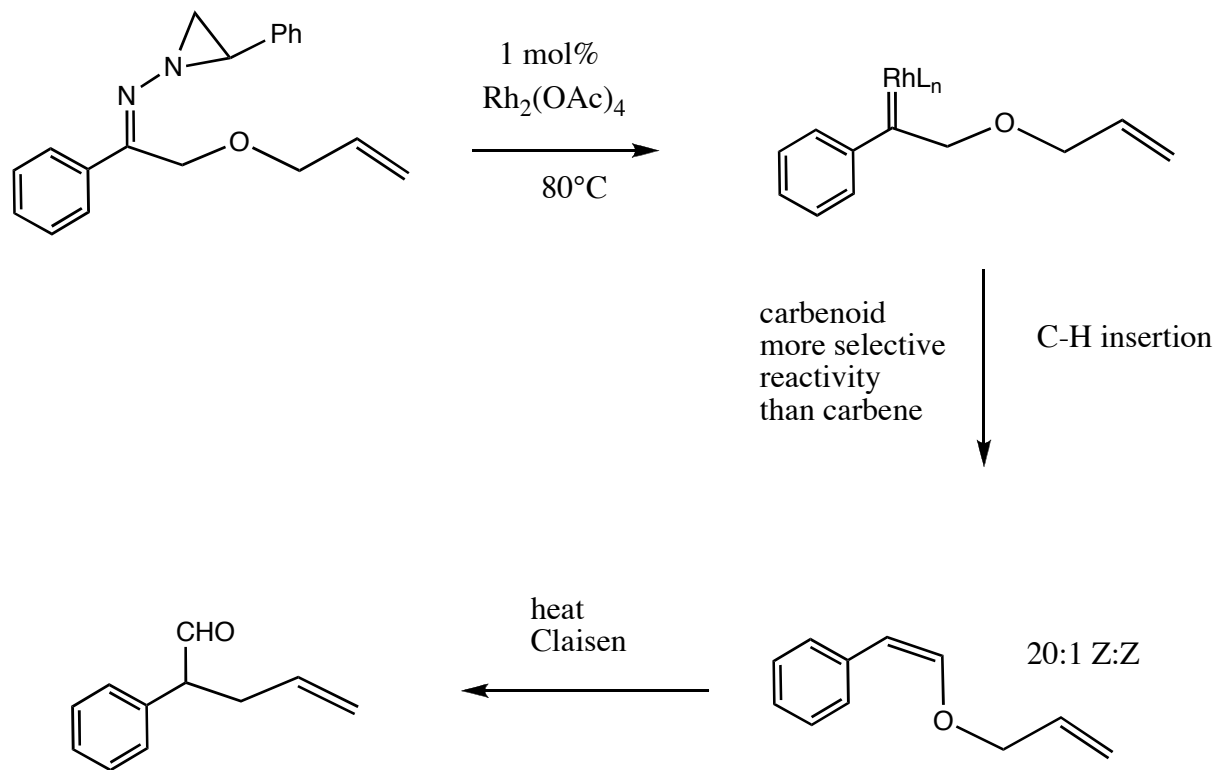
Kinetic Resolution:



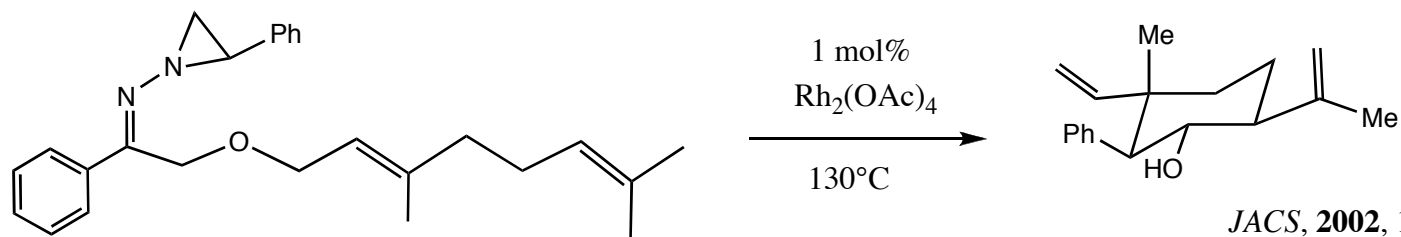
Tandem C-H activation / Cope strategy:



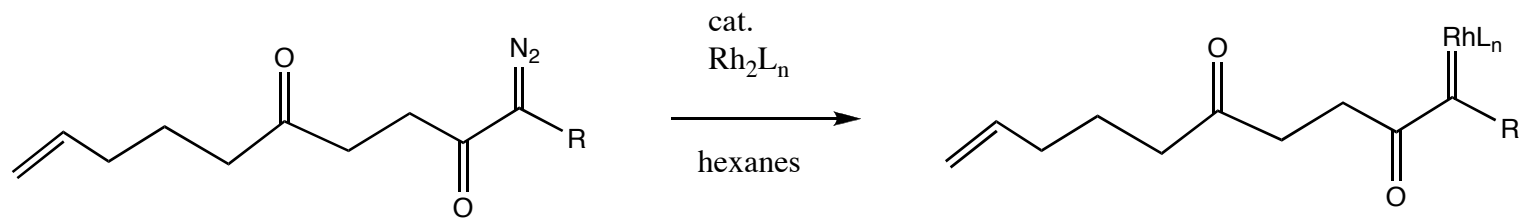
Alternate Carbene/carbenoid processes: Tandem Bamford-Stevens/Claissen sequence



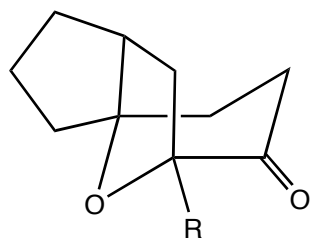
A cascade reaction; explain



Tandem carbenoid formation/ carbonyl ylide formation / 3+2 cycloaddition

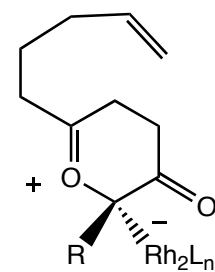


JACS, 2003, 581



up to 90% ee
when chiral ligand used
for Rh_2 species

[3+2] cycloaddition



carbonyl ylide